

**exo-3,6-Dihydro-3-methyl-2-(toluene-4-sulfonyl)-1 $\lambda^4$ ,2-thiazine 1-oxide**Lars Kr. Hansen,<sup>a\*</sup> Annette Bayer<sup>a</sup> and Odd R. Gautun<sup>b</sup><sup>a</sup>Department of Chemistry, University of Tromsø, N-9037 Tromsø, Norway, and<sup>b</sup>Department of Chemistry, Norwegian University of Science and Technology, N-7491 Trondheim, Norway

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**Key indicators**

Single-crystal X-ray study

T = 298 K

Mean  $\sigma(\text{C}-\text{C}) = 0.007 \text{ \AA}$ 

R factor = 0.057

wR factor = 0.162

Data-to-parameter ratio = 10.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $\text{C}_{12}\text{H}_{15}\text{NO}_3\text{S}_2$ , is shown to be the *exo* isomer with a *cis* arrangement of the  $\text{S}=\text{O}$  group and the methyl group in the thiazine ring.

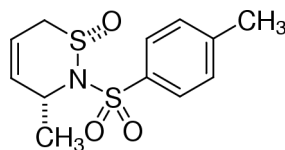
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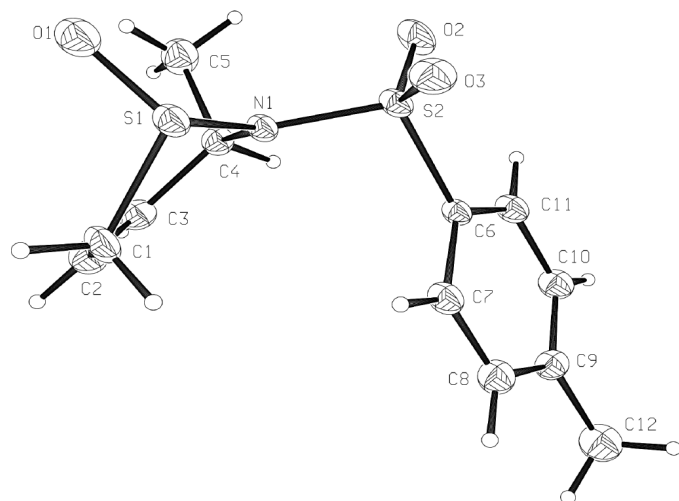
**Comment**

Hetero-[4+2]-cycloadditions of *N*-sulfinyl compounds ( $R-\text{N}=\text{S}=\text{O}$ ) and dienes provide 1,2-thiazine 1-oxides, which can be further transformed into synthetically useful derivatives, e.g. homoallylic amines or vicinal aminoalcohols (Weinreb, 1988). The title compound, (I), was obtained by a [2+4]-cycloaddition of *trans*-1,3-pentadiene and *N*-sulfinyl-4-toluenesulfonamide at room temperature in  $\text{CH}_2\text{Cl}_2$  (Kresze & Wagner, 1972). The reaction product contained two isomers in the ratio 4:1 with different orientations of the  $\text{S}=\text{O}$  bond relative to the methyl group. The *exo* isomer has the  $\text{S}=\text{O}$  bond *cis* to the methyl group, while the *endo* isomer has a *trans* orientation. Here, the structure of the major compound is reported. The major compound was found to be the *exo* isomer.



(I)

The title compound crystallizes in the orthorhombic non-centrosymmetric space group  $P2_12_12_1$  and is therefore a conglomerate under the applied conditions of crystallization. A molecule with the atomic numbering scheme is shown in Fig. 1. The thiazine part of the molecule is found to be in a half-chair conformation, and one also sees that the  $\text{S}=\text{O}$  and the C4–C5 bonds are on the same side of the thiazine ring, giving the *exo* isomer. Least-squares planes through the phenyl part (atoms C6–C11) and the thiazine part (atoms C1–C4) show an angle of  $71.2(2)^\circ$  between the two planes. Atoms S1 and N1 are displaced by  $0.525(9)$  and  $0.261(8)$  Å, respectively, on opposite sides with respect to the plane through atoms C1–C4. The total puckering amplitude parameter (Cremer & Pople, 1982)  $Q_T = 0.499(3)$  Å and one notices the presence of a local twofold axis running through the midpoints of the S1–N1 and C2–C3 bonds. Atoms C12 and S2 nearly lie in the phenyl group plane, being displaced by  $0.016(9)$  and  $0.099(6)$  Å on the same side with respect to that plane. A selection of bond lengths shows that these are all within the normal range of such bonds (Allen *et al.*, 1987).



**Figure 1**

A view of the title compound with the atomic numbering scheme. Displacement ellipsoids are drawn at the 20% probability level.

$S2=O2$  and  $S2=O3$  are both 1.428 (3) Å, while the  $S1=O1$  distance is 1.473 (4) Å. The molecules are packed in the crystal through a series of intra- and intermolecular short contacts (Taylor & Kennard, 1982) (see Table 1).

## Experimental

The 1,4-thiazine 1-oxide was dissolved in  $CH_2Cl_2$  and heptane was added until saturation. Crystals were grown by vapour diffusion of the solvent in the refrigerator.

### Crystal data

$C_{12}H_{15}NO_3S_2$   
 $M_r = 285.37$   
 Orthorhombic,  $P2_12_12_1$   
 $a = 8.277$  (2) Å  
 $b = 9.723$  (2) Å  
 $c = 16.500$  (2) Å  
 $V = 1327.9$  (5) Å<sup>3</sup>  
 $Z = 4$   
 $D_x = 1.427$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation  
 Cell parameters from 25 reflections  
 $\theta = 8-14^\circ$   
 $\mu = 0.40$  mm<sup>-1</sup>  
 $T = 298$  (2) K  
 Block, colourless  
 $0.5 \times 0.3 \times 0.2$  mm

### Data collection

Enraf–Nonius CAD-4 diffractometer  
 $\omega-2\theta$  scans  
 1710 measured reflections  
 1669 independent reflections  
 1386 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.020$

$\theta_{max} = 26.9^\circ$   
 $h = 0 \rightarrow 10$   
 $k = -3 \rightarrow 12$   
 $l = -6 \rightarrow 21$   
 3 standard reflections  
 frequency: 120 min  
 intensity decay: 1%

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.057$   
 $wR(F^2) = 0.162$   
 $S = 1.06$   
 1669 reflections  
 165 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1332P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} < 0.001$   
 $\Delta\rho_{max} = 0.91$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.41$  e Å<sup>-3</sup>  
 Absolute structure: Flack (1983)  
 Flack parameter = 0.12 (19)

**Table 1**

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C1-H1B \cdots O2^i$	0.97	2.39	3.225 (6)	144
$C11-H11 \cdots O2$	0.93	2.51	2.888 (6)	104
$C3-H3 \cdots O2^{ii}$	0.93	2.58	3.487 (6)	165
$C1-H1A \cdots O3^{iii}$	0.97	2.66	3.601 (6)	164
$C5-H5A \cdots O1$	0.96	2.56	3.249 (8)	128
$C2-H2 \cdots O1^{iv}$	0.93	2.63	3.555 (6)	178

Symmetry codes: (i)  $1+x, y, z$ ; (ii)  $1-x, \frac{1}{2}+y, \frac{1}{2}-z$ ; (iii)  $\frac{1}{2}+x, \frac{3}{2}-y, 1-z$ ; (iv)  $2-x, \frac{1}{2}+y, \frac{1}{2}-z$ .

Data collection: *CAD-4-PC Software* (Enraf–Nonius, 1992); cell refinement: *CELDIM* in *CAD-4-PC Software*; data reduction: *XCAD* (McArdle & Higgins, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEX* (McArdle, 1995); software used to prepare material for publication: *OSCAIL* (McArdle, 1993).

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